

THE EFFECT OF PARTICLE SIZE ON THE SUBLIMATION OF BUTYLHYDROXYTOLUENE IN TABLETS

M. Decorte¹, B. Van Hove², F. Preda², E. Verheyen², N. Moazami Goudarzi³, M. Boone³, C. Vervaet¹, V. Vanhoorne¹

¹ Ghent University, Laboratory of Pharmaceutical Technology, Ottergemsesteenweg 460, 9000 Ghent, Belgium

² Janssen, Pharmaceutical companies of Johnson & Johnson, division Chemical and Pharmaceutical Development and Supply, Turnhoutsesteenweg 30, 2340 Beerse, Belgium

³ Ghent University, Centre for X-Ray Tomography – UGCT, Proeftuinstraat 86, 9000 Ghent, Belgium

1. INTRODUCTION

- An increase in the number of oxidation-sensitive APIs in recent years can require the **implementation of antioxidants in oral solid dosage forms**.
- The antioxidant butylhydroxytoluene (BHT) **sublimates at higher temperatures**, which could cause problems during processing.

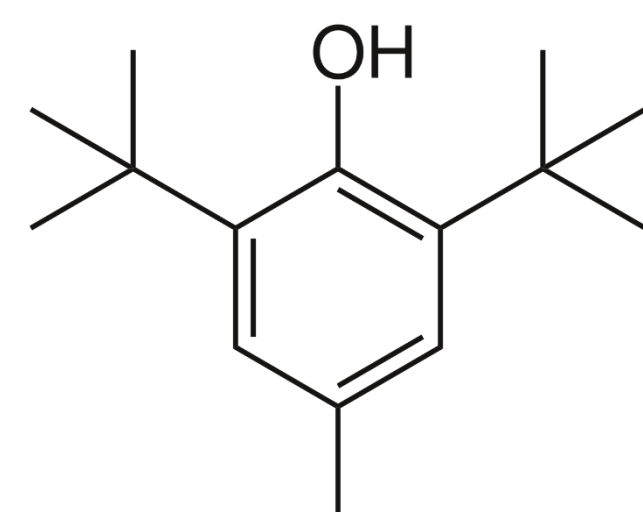


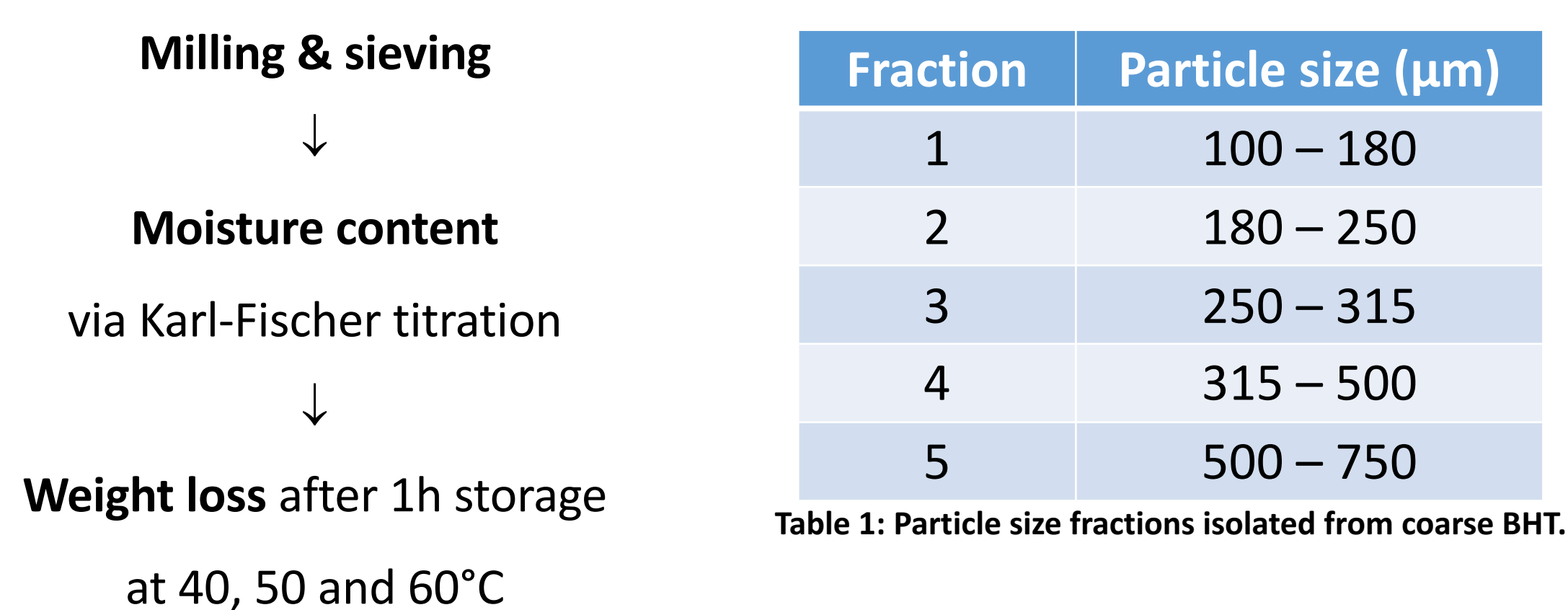
Figure 1: Structure of BHT.

2. OBJECTIVES

- Determination of the extent of BHT sublimation **in tablets**.
- Investigating the **influence of particle size** of BHT on the sublimation behaviour.

3. MATERIALS & METHODS

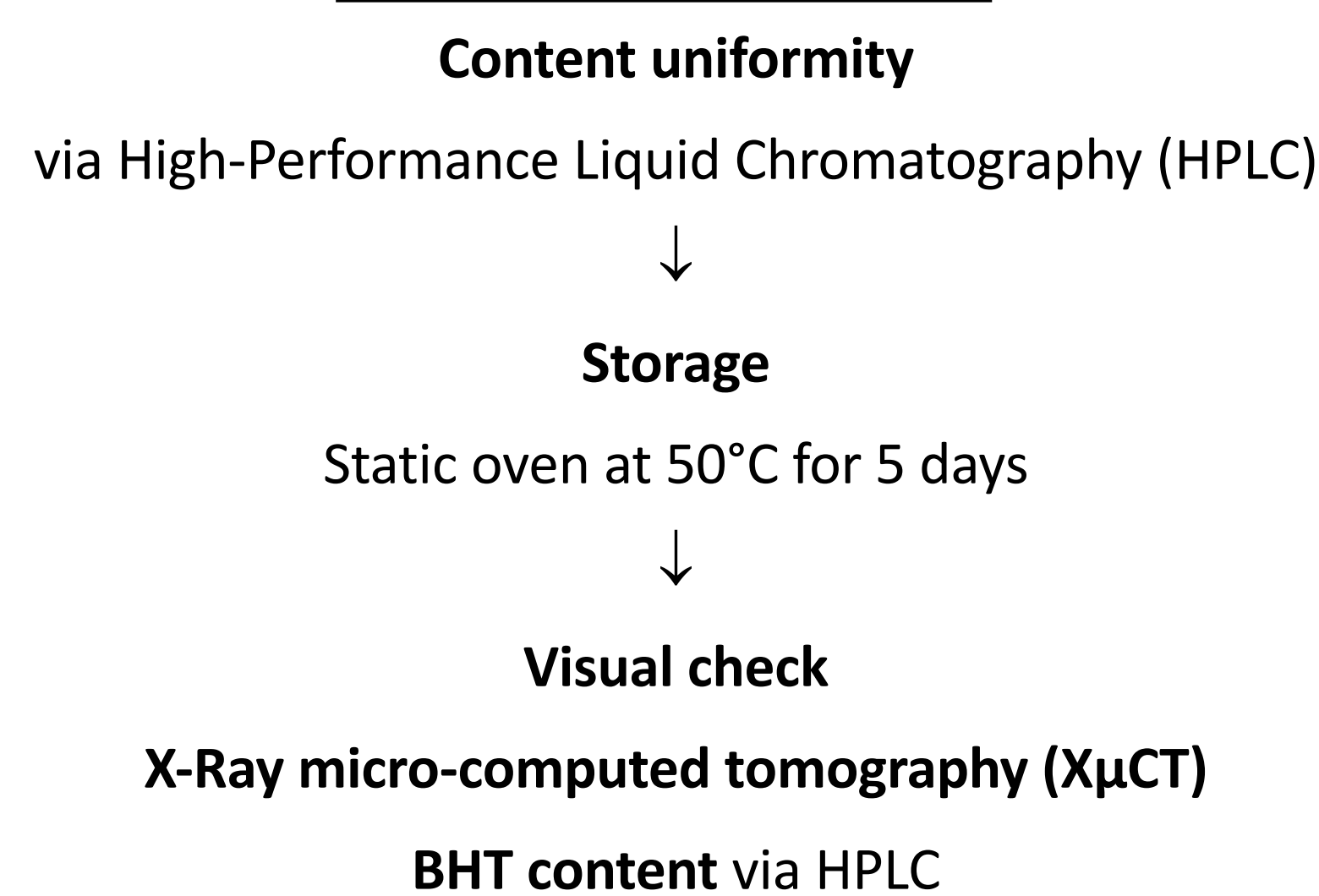
Fractionation of BHT



Tableting

- MedelPharm STYL'One Evo**
→ Uniaxial confined compression at 25 kN in a 11.28 mm die
- Formulation:** 1% w/w BHT – 0.5% w/w MgSt – 98.5% w/w MCC PH-102

Sublimation in tablets



5. CONCLUSION

- No relationship between particle size and degree of sublimation of pure BHT.
- Inclusion of BHT in a tablet did not prevent sublimation.
- Particle size influenced the sublimation rate.** Larger particles sublime slower, as some BHT was left in core of tablet when particle size > 250 μm.

The agency Flanders Innovation & Entrepreneurship and Janssen Pharmaceutica are acknowledged for co-funding of this project. The financial support of Ghent University is also acknowledged for the XμCT measurements.

Contact: Milan.Decorte@UGent.be



4. RESULTS

Sublimation of pure BHT

Moisture content:

- For all BHT size fractions <0.14%.
- No correlation between size fraction and moisture content.
- All weight loss during storage can be attributed to BHT sublimation.

Weight loss:

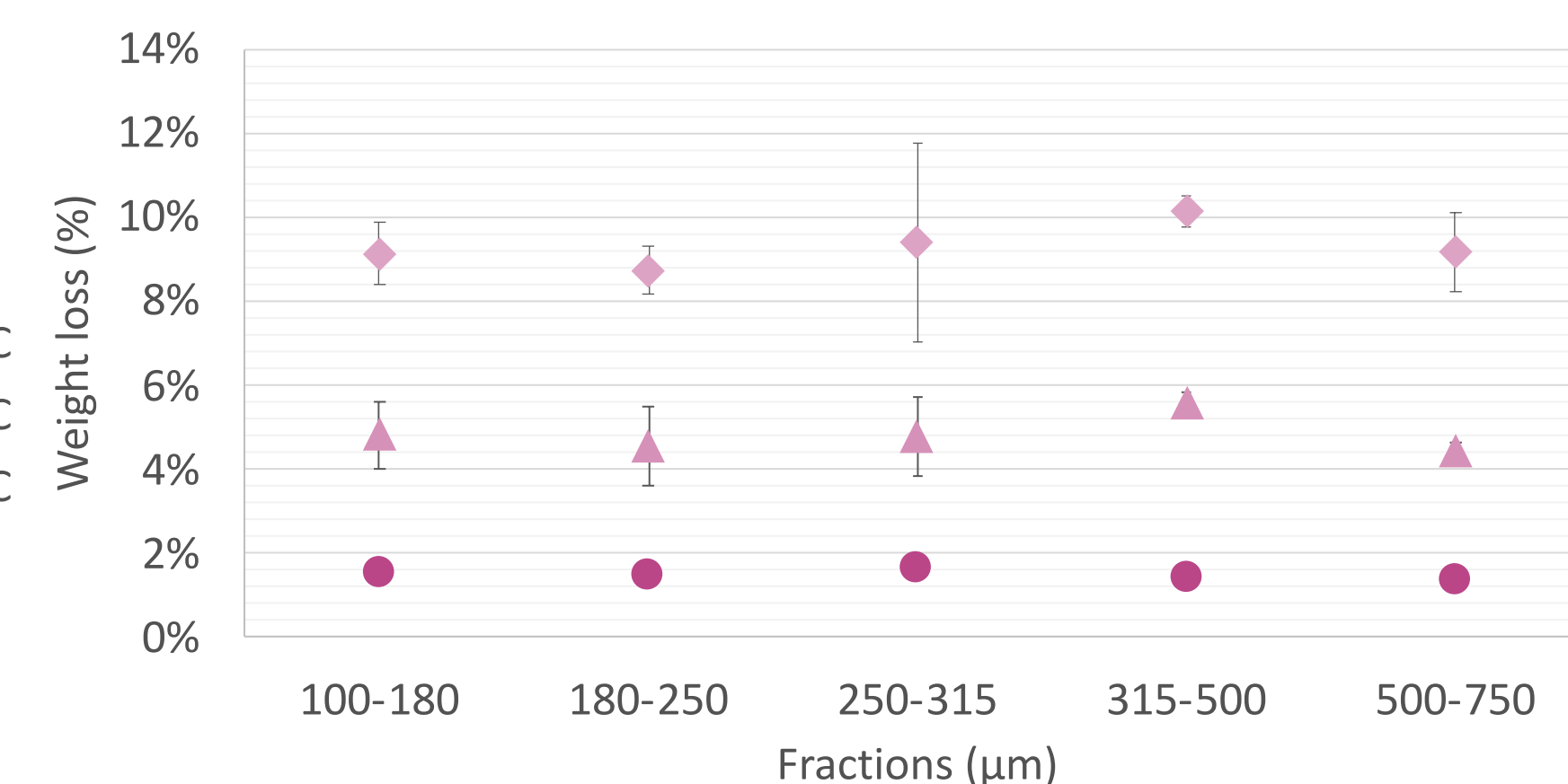


Figure 2: Weight loss of the BHT size fractions stored for 1 hour at different temperatures.

- Sublimation degree ~ temperature.
- No significant difference between size fractions of BHT.
- Partial melting at 60°C. (T_m BHT: 70°C)

Sublimation of BHT in tablets

Content uniformity:

- Fraction 1-4:** 100.3 ± 7.3 %
→ good content uniformity.
- Fraction 5:** 78.7 ± 34.0 %
→ inhomogeneous BHT distribution.
- Using larger particle size
→ Higher variability of BHT content.

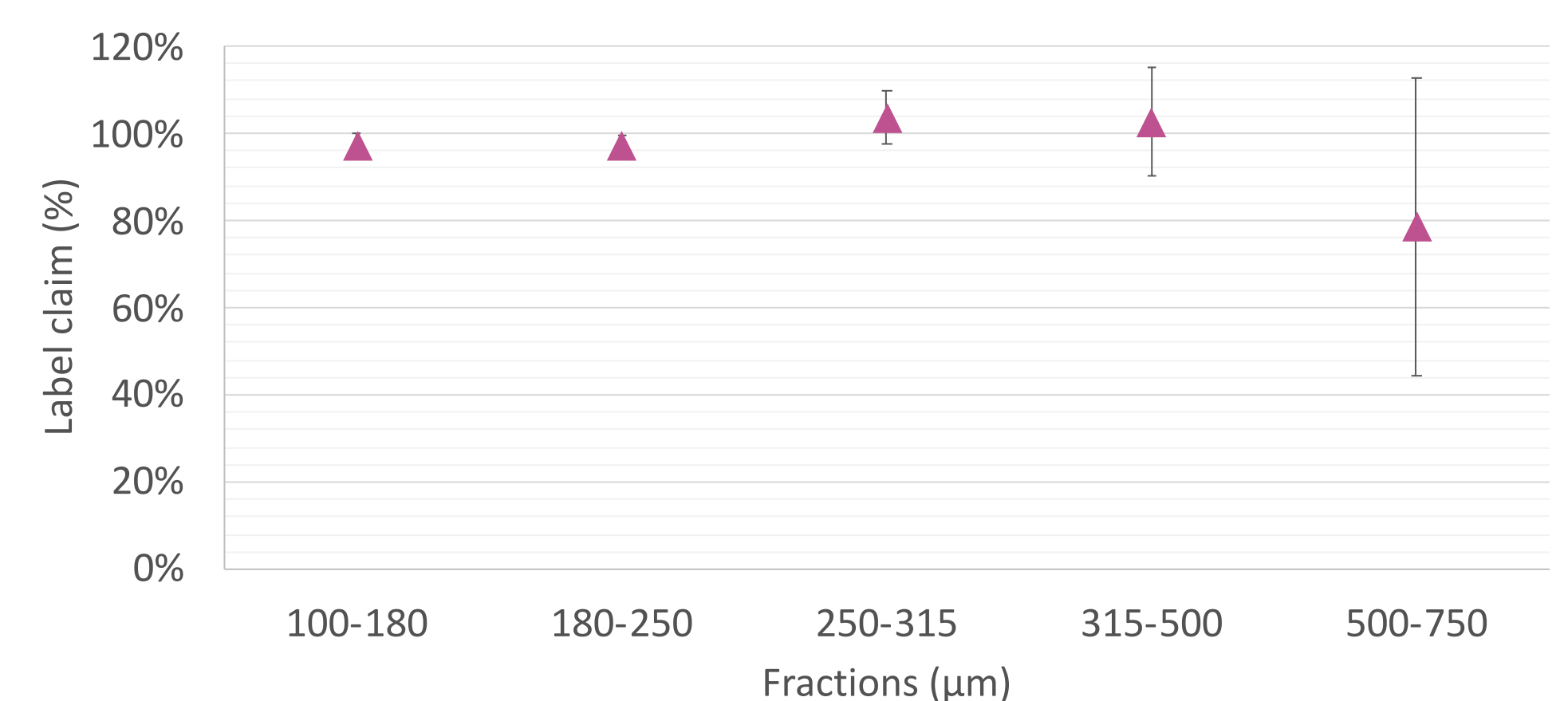


Figure 3: Content uniformity of tablets prepared with each size fraction of BHT.

Sublimation degree:

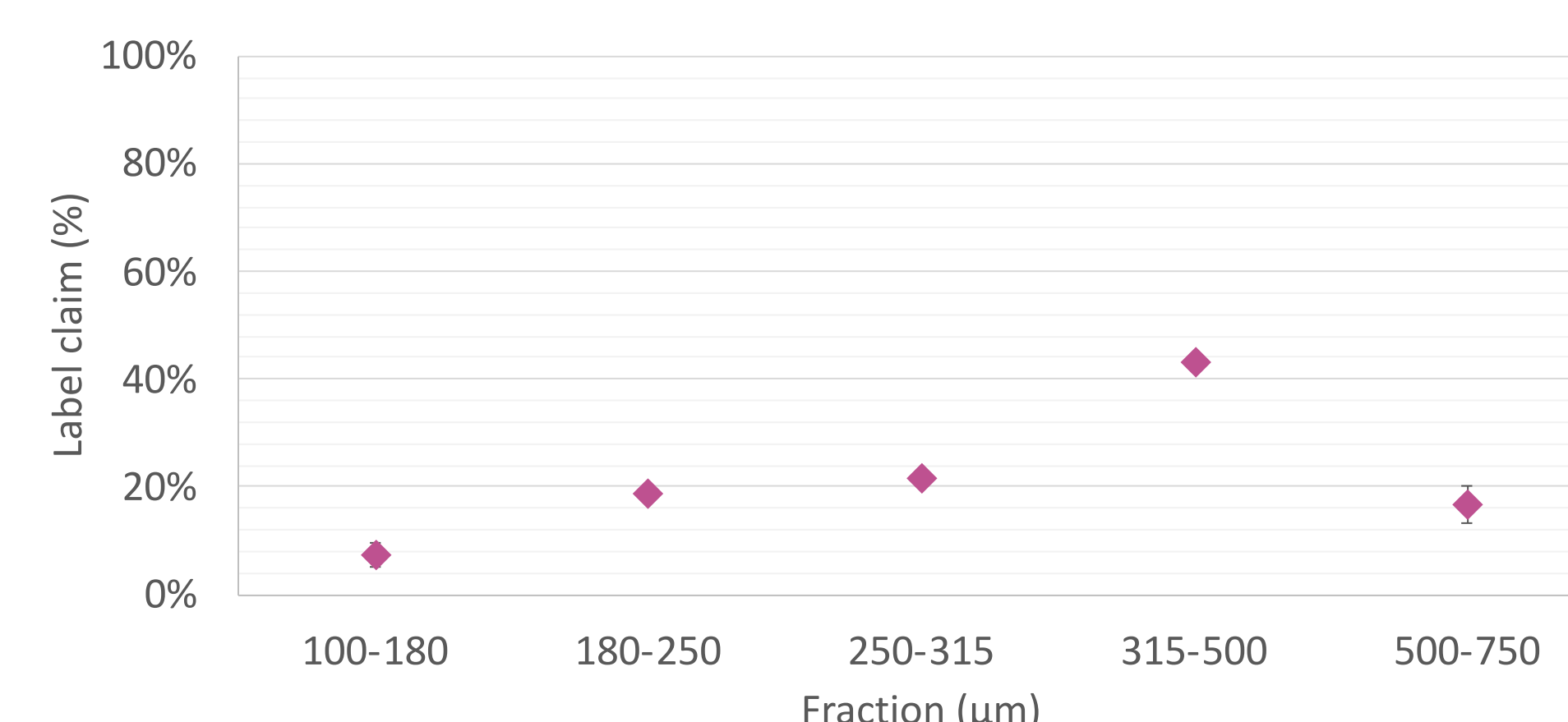


Figure 4: BHT content in tablets after 5 days storage at 50°C.

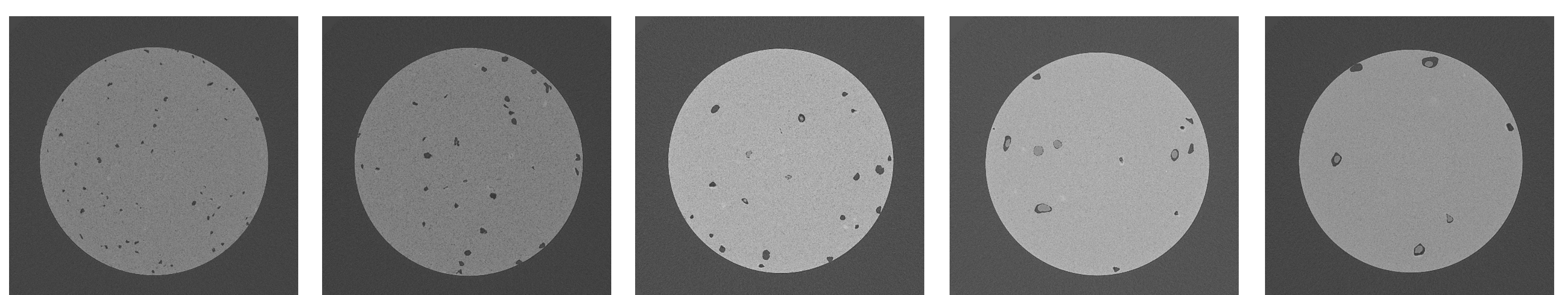
- Residual BHT ~ particle size.
- Using larger particle size
→ lower specific surface area
→ sublimation ↓
- Fraction 4:** 43.3 ± 1.2% BHT remaining.
- Fraction 5:** unreliable data due to content uniformity issues.

Pore formation:



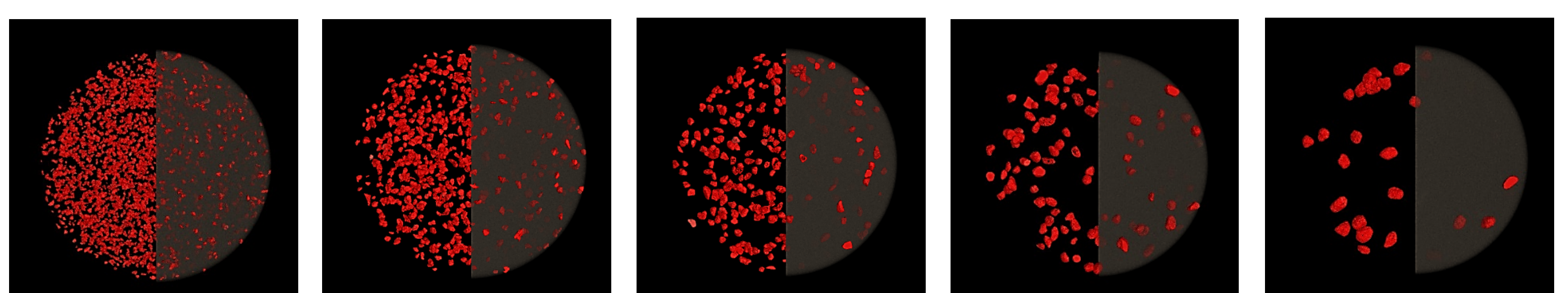
- Appearance:**
- Pores observed at tablet surface for all fractions.
 - Pore size ~ particle size.

Figure 5: Top view of a heat-treated tablet containing different BHT size fractions. From left to right: fraction 1 to 5.



- XμCT scans:**
- Pore formation both at surface and in core of tablet.
 - Using larger particle size → BHT residue visible on XμCT.

Figure 6: XμCT scan of the middle section of a heat-treated tablet containing different BHT size fractions. From left to right: fraction 1 to 5.



- 3D-reconstruction:**
- Red spots = pores without BHT residue.
 - Slower sublimation rate of BHT in the core of the tablet.

Figure 7: 3D-reconstruction of a heat-treated tablet containing different BHT size fractions. From left to right: fraction 1 to 5.